# Radiation Chemistry vs. Photochemistry Cosmic Synthesis of Prebiotic Molecules

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First Evidence of molecules in space: Annie Jump Cannon (Wellesley College Class of 1884)

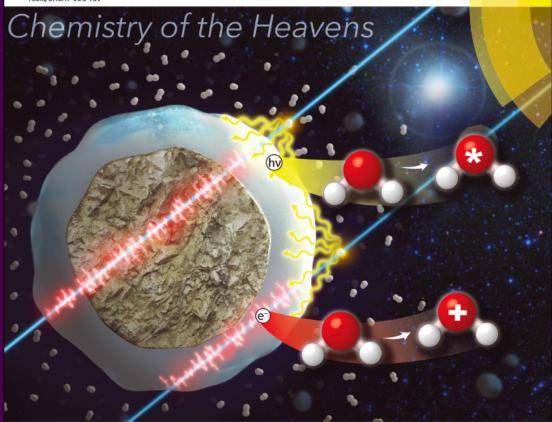
#### 1. Introduction

A long-standing question in astronomical spectroscopy has been the identity of the carriers of the diffuse interstellar bands (DIBs). DIBs are broad absorption features, primarily in the visible region, that have been observed in the spectra of numerous stars [1, 2]. After the incidental observation of the strongest DIB at 4430 Å (henceforth referred to as the  $\lambda4430$  DIB) by Annie Jump Cannon in the early years of stellar spectroscopy [3–5] and reports of the  $\lambda\lambda5780$  and 5797 DIBs by Mary Heger in 1922 [6, 7], Paul W. Merrill demonstrated in the 1930s that these spectral features arose due to intervening interstellar material [8, 9]. Laboratory spectroscopy and computational modeling have led to a diverse set of hypotheses for the origins of DIBs. Douglas argued for the consideration of long-

Molecular Physics, **2015**,113, 2159-2168

### **Chem Soc Rev**

Chemical Society Reviews rsc.li/chem-soc-rev



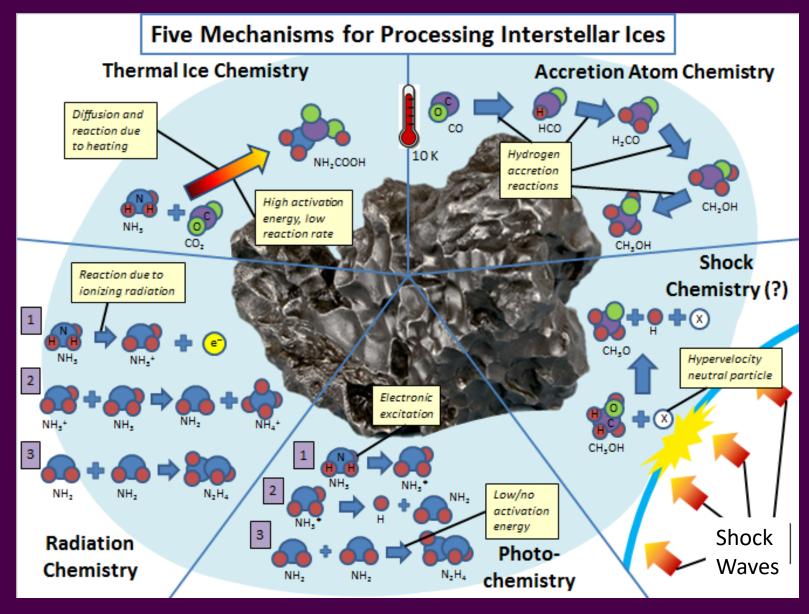
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#### TUTORIAL REVIEW

Chris R. Arumainayagam et al. Extraterrestrial prebiotic molecules: photochemistry vs. radiation chemistry of interstellar ices

#### Five Mechanisms for Processing Interstellar Ices



Arumainayagam et. al., Chemical Society Reviews 2019, 48, 8, 2267–2496.

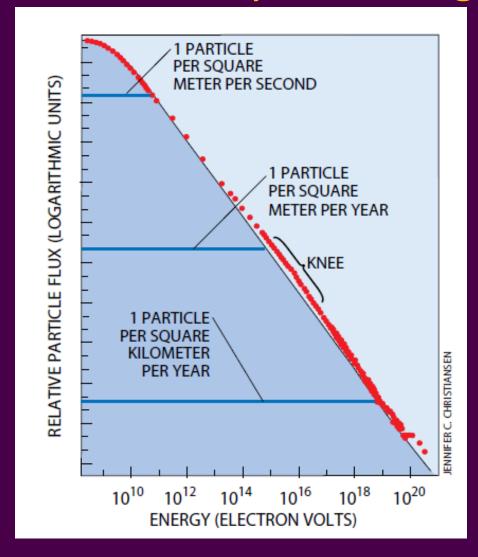
## What is the difference between radiation chemistry and photochemistry?

### **Radiation Chemistry**

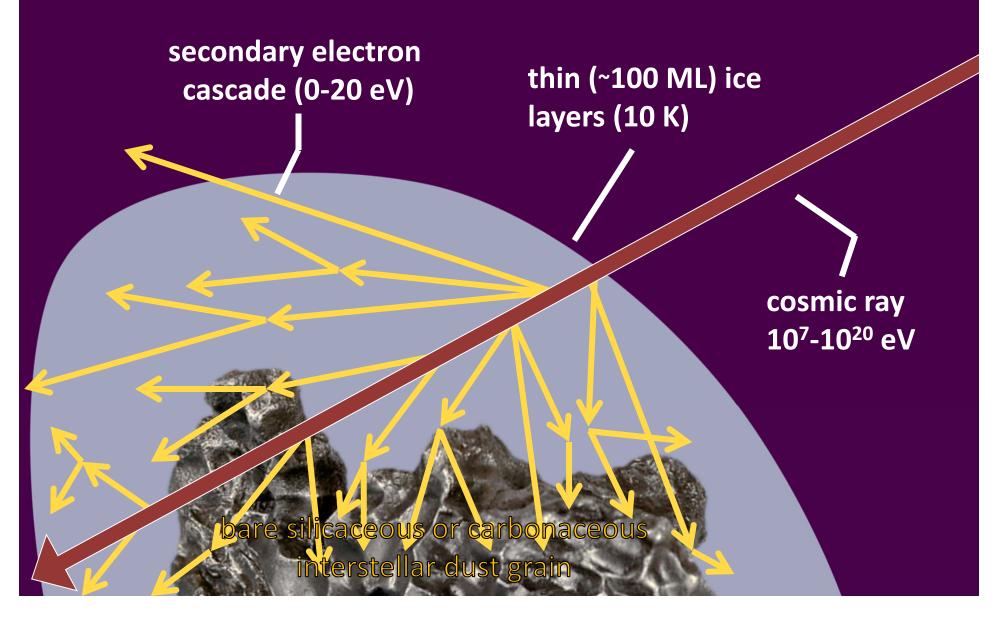
Radiation chemistry is the "study of the chemical changes produced by the absorption of radiation of sufficiently high energy to produce ionization."

RADIATION CHEMISTRY INVOLVES IONIZATION.

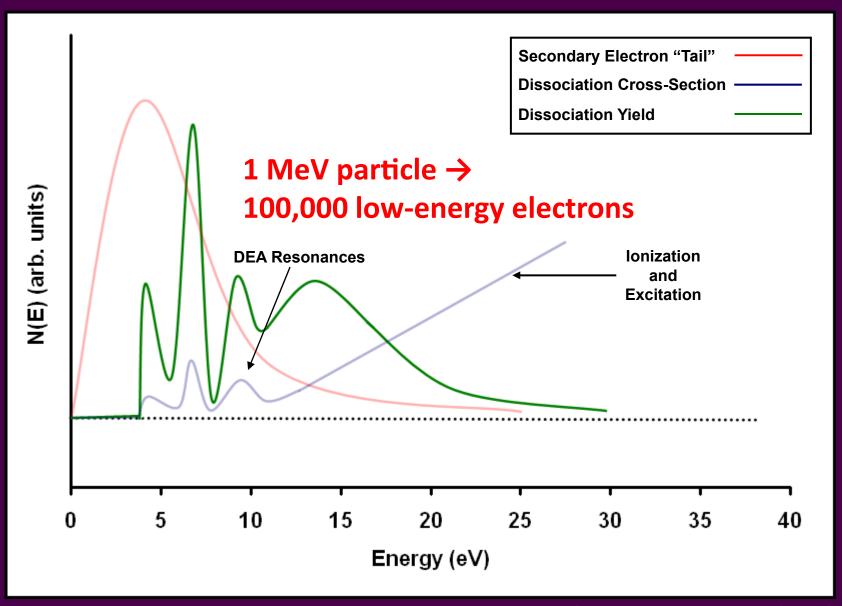
#### Flux of Cosmic Rays Reaching Earth



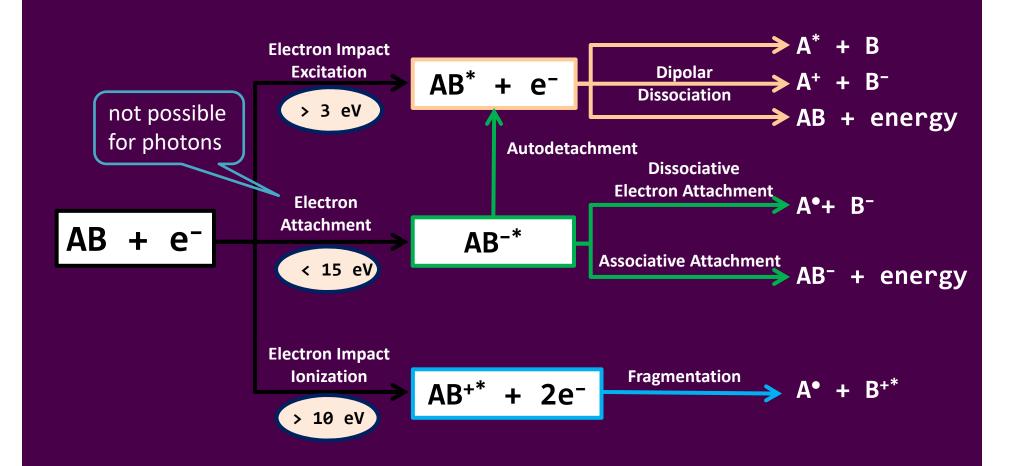
# Formation of Secondary Electrons in Cosmic Ices and Dust Grains



### Importance of Low-Energy Electrons



#### Electron-induced Dissociation Mechanisms



#### Photochemistry

Photochemistry involves "chemical processes which occur from the electronically excited state formed by photon absorption."

- visible (1.8 eV 3.1 eV)
- near-UV (3.1 4.1 eV)
- far (deep)-UV (4.1 6.2 eV)
- vacuum-UV (6.2 -12.4 eV)

no ionization; only excitation

ionization AND excitation

### Condensed Phase Ionization Energy

 $IE_{condensed} < IE_{gas}$  phase phase

Most previous

"photochemistry"

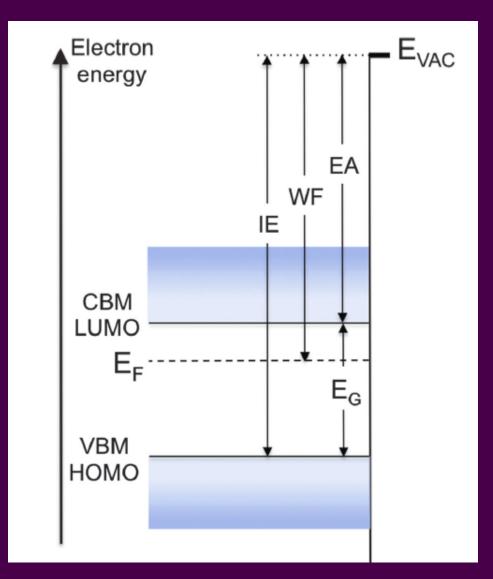
studies relevant to

astrochemistry

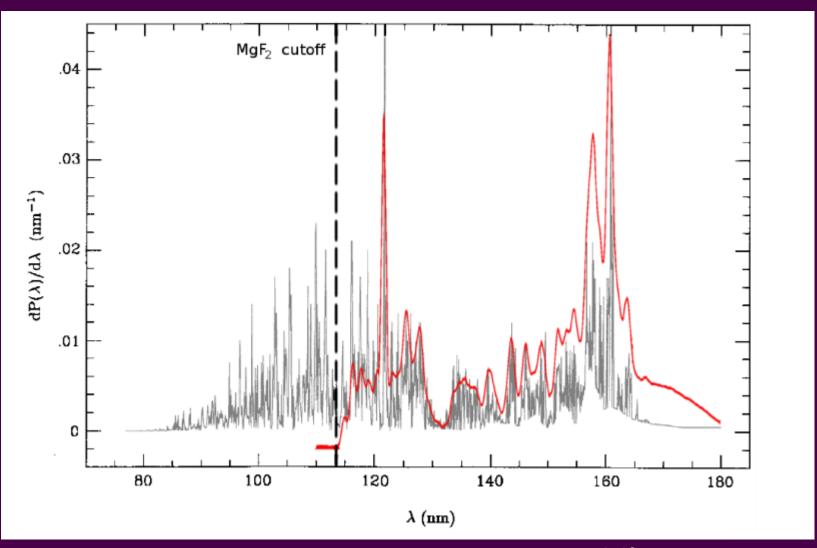
involve radiation

chemistry in addition

to photochemistry.

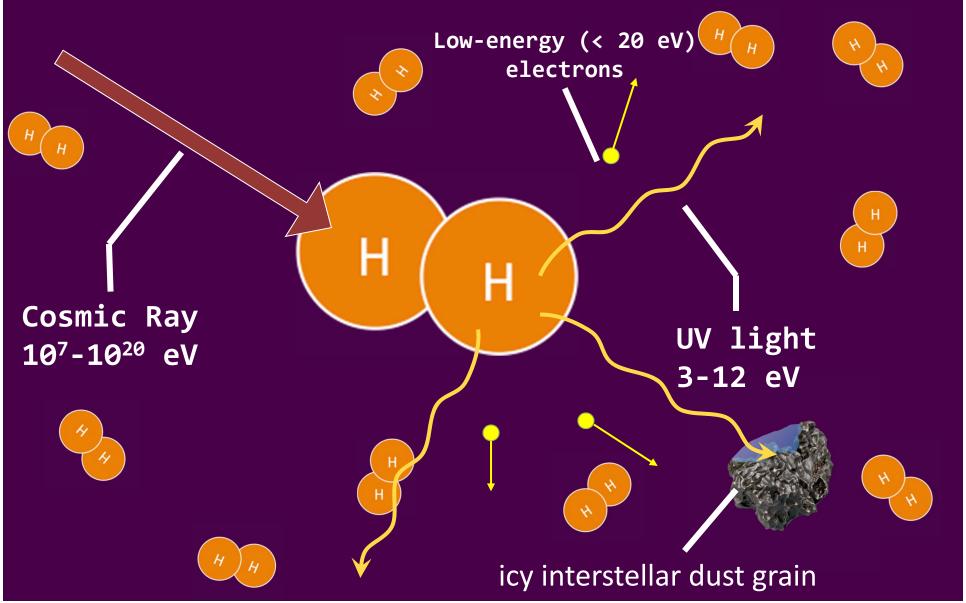


### VUV-emission spectrum of Microwave-Discharge Hydrogen-Flow Lamps (MDHL)

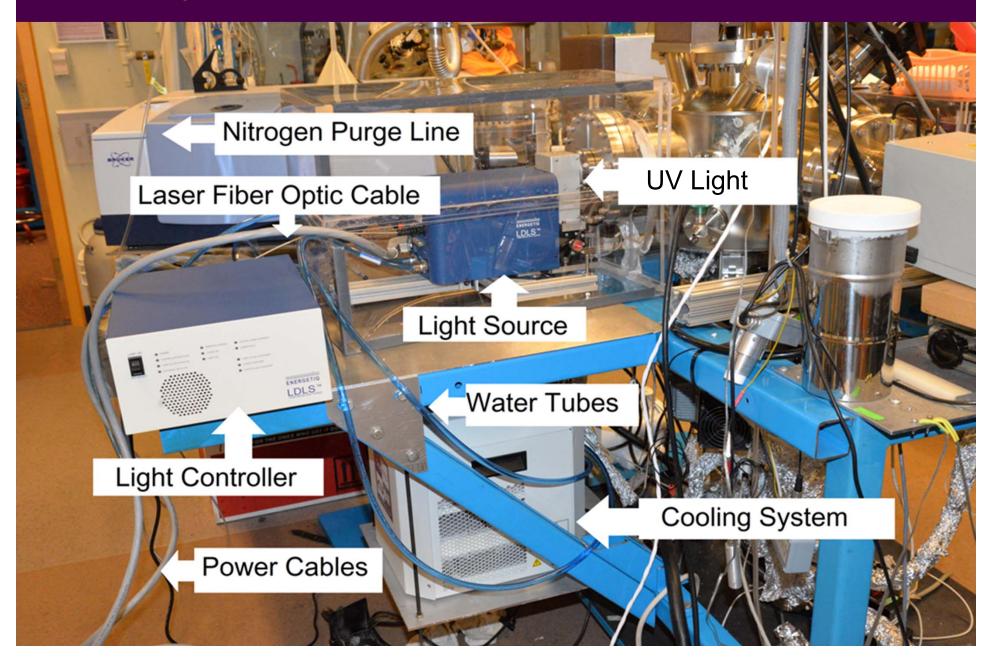


**CREDIT: Gustavo Adolfo Cruz Diaz** 

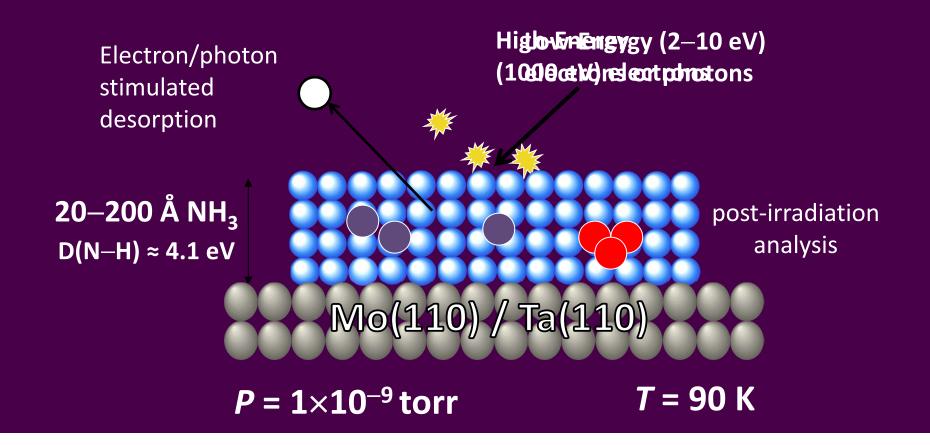
# UV light formation within dark, dense molecular clouds

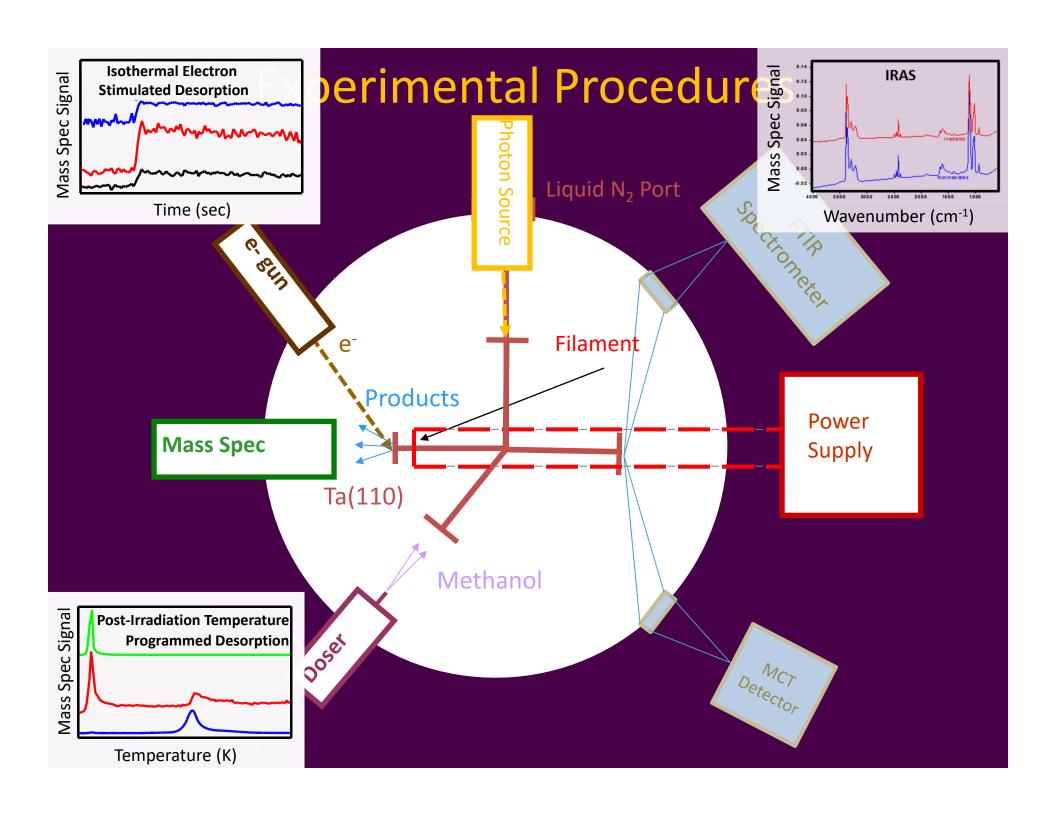


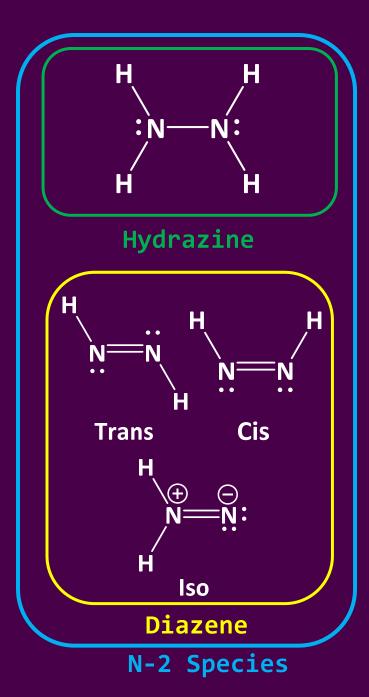
#### How do you create a little bit of heaven on earth?

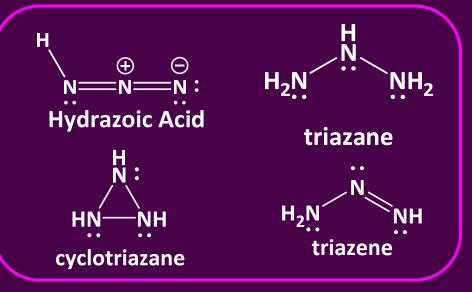


### **Experimental Procedures**



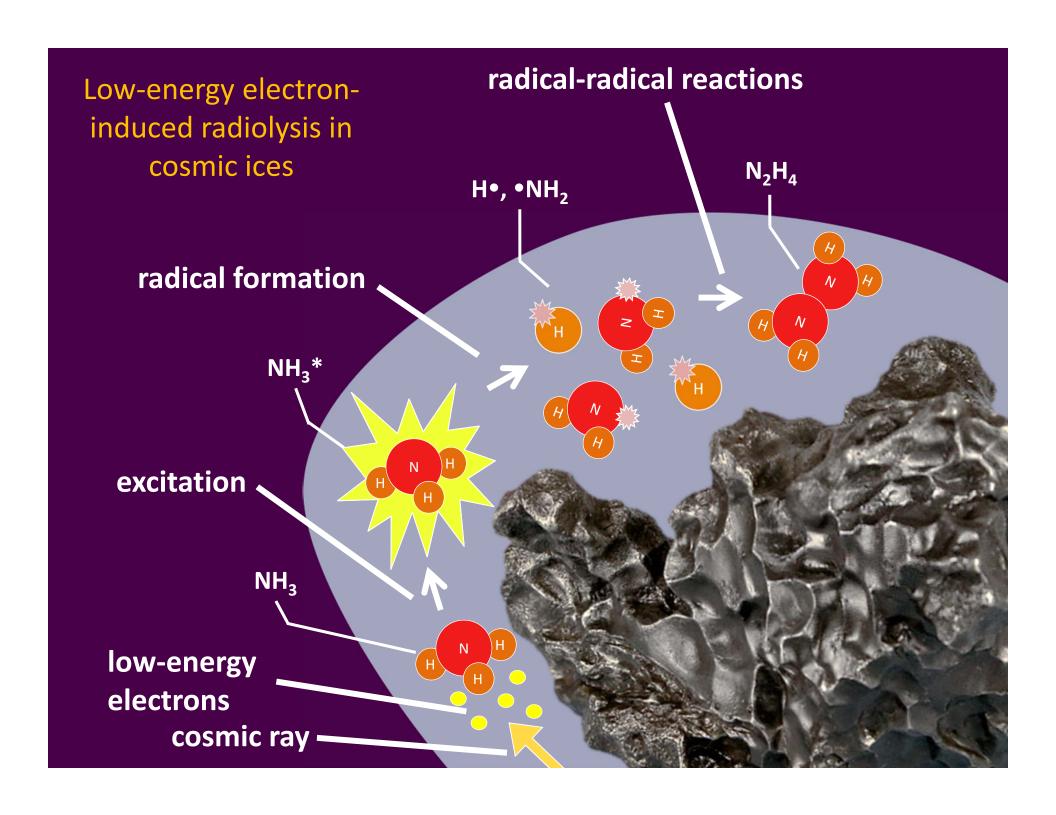




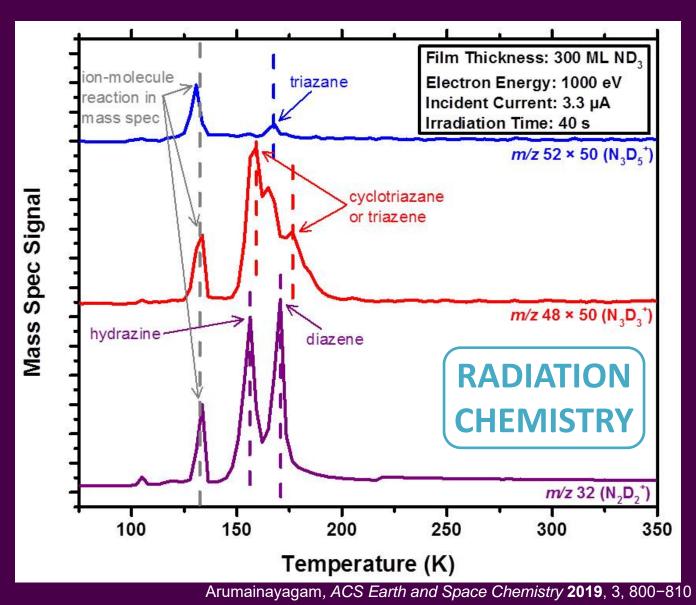


N-3 Species

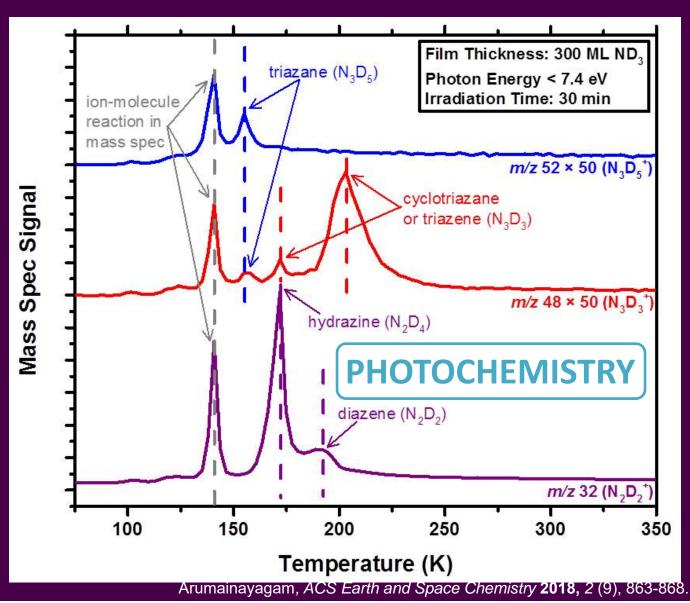
Possible Radiolysis/Photolysis Products of Ammonia



### Products of High Energy Electrons From Condensed Ammonia



# Products of Low-Energy Photons From Condensed Ammonia



### **Energetics of Ammonia Photolysis**

- Ionization energy for gas phase ammonia: 10.1 eV
- Threshold for low-energy secondary electrons: 8.5 eV
- Absorption threshold photon energy: 6.2 eV

Our experiments were done at < 7.4 eV.

Most previous "photochemistry" studies relevant to astrochemistry involve radiation chemistry in addition to photochemistry.

Our experiments involve only photochemistry.

#### Conclusions

 Low-energy (< 20 eV) electrons may be the dominant species responsible for the synthesis of prebiotic molecules in the interstellar medium

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